

T h e P h a n t o m L a b o r a t o r y

RSVP Phantom™ Pelvis M a n u a l

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Caution

Federal law restricts this device to sale by or on the order of a physician, or with the descriptive designation of any other practitioner licensed by the law of the State in which he/she practices to use or order the use of the device.

This product is intended for use by medical physicists with training in radiation therapy. A medical physicist must design and supervise the actual tests performed with this phantom. This manual does not provide information regarding applications of the product only information regarding the product functions.

WARNING

This product has an FH3-4 mm/min flame rating and is considered to be flammable. It is advised not to expose this product to open flame or high temperature (over 125° Celsius or 250° Fahrenheit) heating elements.

TLP260

6/27/11

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RSVP Phantom™ Pelvis Manual

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Indications for Use

The RSVP Phantom™ Pelvis is designed for use in a variety of radiation therapy applications including, final quality verification of therapy dose delivery and for comparing the delivered dose profiles for different treatment plans. It is also used for periodic quality assurance evaluations and acceptance testing and to perform re-evaluations after equipment or software upgrades.

Introduction

The Phantom Laboratory, Incorporated and physicist Charles W. Coffey, II, Ph.D. have developed the RSVP Phantom™ Pelvis to provide localization and dose verification for IMRT systems.

A variety of techniques may be used to measure the radiation dose delivered to the tumor vessel. Quantitative dose measurements may also be obtained by using commercially available probes and ionization chambers.

Because each medical facility has its own unique set of requirements, we do not make specific recommendations on the frequency and expected results of evaluations. When developing your measurement protocols, review the local governing regulations, the needs of your radiologists, oncologists, physicists, and your system manufacturer's recommendations.

If you have questions about the operation of the RSVP Phantom™ Pelvis or the optional equipment that is available for the phantom, please contact The Phantom Laboratory at the following address:

The Phantom Laboratory, Incorporated
PO Box 511, Salem NY 12865 USA
Phone: 800-525-1190 or 518-692-1190
Fax: 518-692-3329

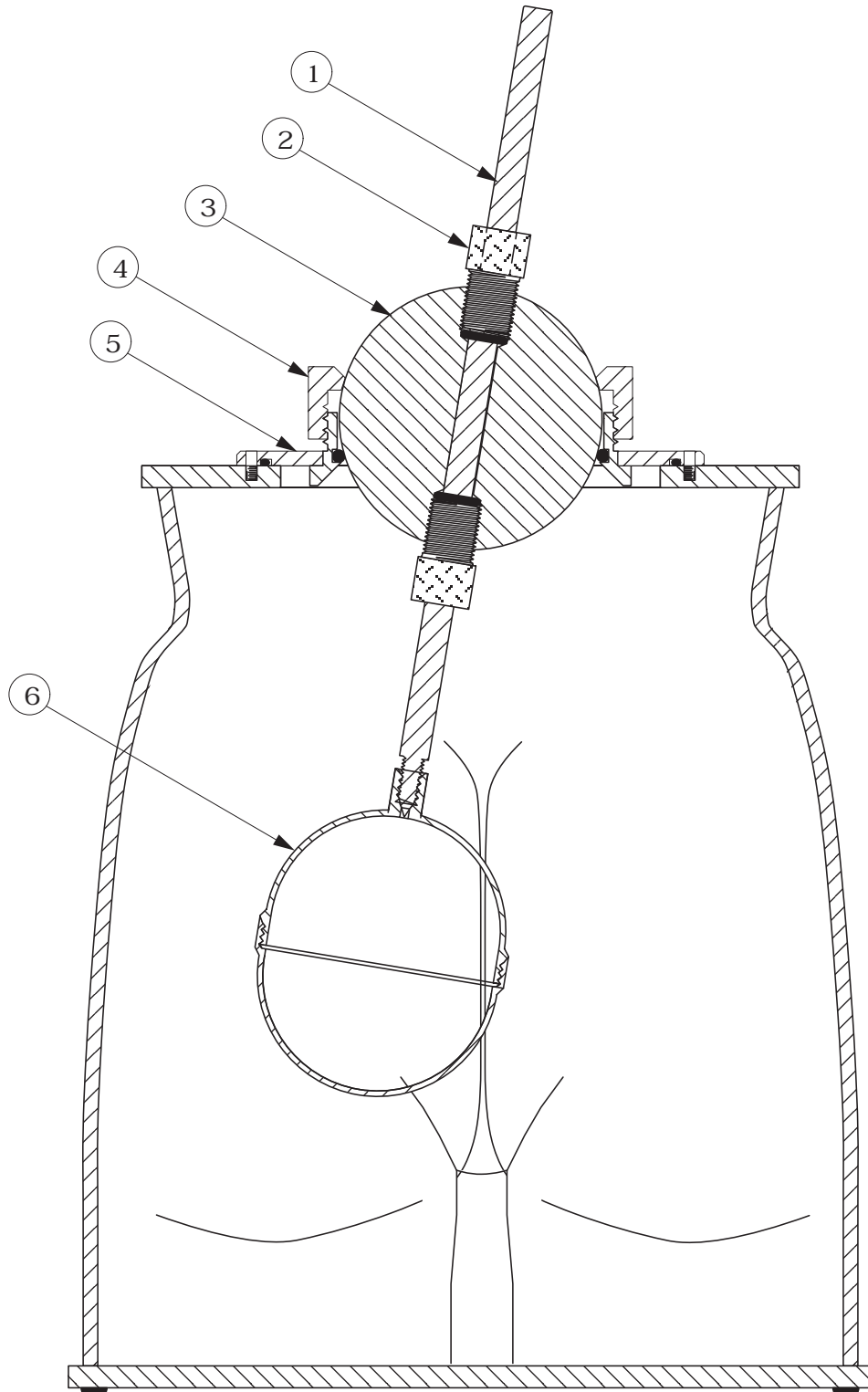


Figure 1

Cross section view of the RSVP Phantom™ Pelvis

1. Tumor support rod
2. Lock bolt
3. Rotation ball
4. Rotation ball lock nut
5. Cover plate
6. Tumor vessel

Care of the Phantom

The shell of the RSVP Pelvis Phantom™ is formed from 0.25-inch-thick cellulose acetate butyrate (CAB), a transparent plastic chosen for its strength and low water absorption. The shell is mounted on durable polycarbonate end plates. A cover plate and an assembly, that allows the tumor port to be rotated, is attached to the top.

The RSVP Phantom™ is designed to be filled with water. Please refer to the Chemical Compatibility Chart on pages 8-13 of this manual prior to use. You should never allow a filled phantom to freeze. If water is left in the phantom for an extended period of time, bacteria and other microorganisms may begin to grow and cause the water to become cloudy. If clouding occurs, the phantom should be emptied and washed with water and a mild detergent. Before storing, empty all water from, the phantom.

The tumor rotation ball assembly is located on the top of the phantom. The ball assembly may be extracted from the phantom by simply removing its lock nut and then lifting the ball out of the assembly. The O-ring that lies under the rotation ball should sit evenly on the lip surrounding the opening that leads into the main body of the phantom. The application of a small amount of petroleum jelly to the O-ring will help insure that the rotation ball forms a tight seal on the ring. Be careful not to apply too much jelly to the O-ring, because excessive lubrication may cause the ring to slip out of position when the lock nut is tightened. Always remember to loosen the rotation ball lock nut before rotating the tumor ball. If the rotation ball is turned when the nut is tight, the O-ring will be twisted out of position, thereby preventing a proper seal.

Tumor Removal

To remove the tumor vessel, remove the rotation ball lock nut. Once this is accomplished, the rotation ball, tumor support rod, and tumor vessel can be lifted out of the phantom as a complete assembly, as shown in Figure 2.

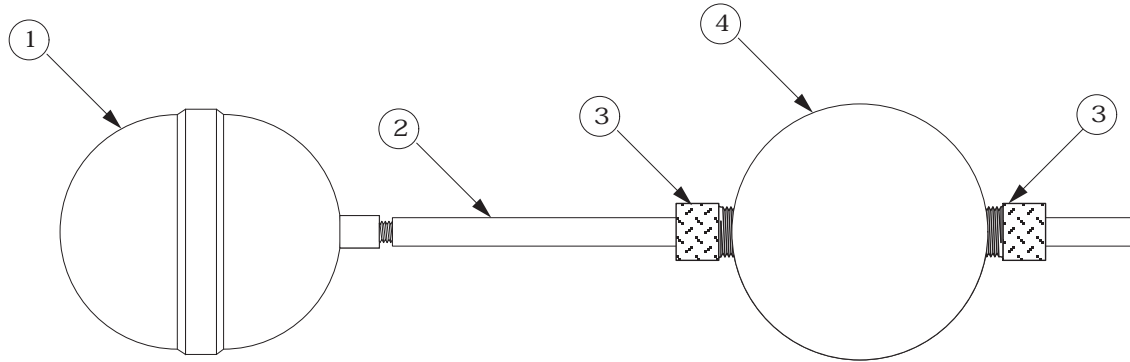


Figure 2

1. Tumor vessel
2. Tumor support rod
3. Lock bolt
4. Rotation ball

Chemical Compatibility Chart

Reagent Chemicals	Time Exposed	Percent Increase Weight Thickness		Observed Condition of Plastic
Acids				
Acetic, 5%	1 year	3.6	2.1	Slightly softened
Acetic, 10%	2 months	5.2	2.4	Slightly softened
•Acetic, 30%	2 months	13.6	8.6	Softened and swollen
Boric, 5%	2 days	1.3	0.0	Unchanged
Chromic, 6%	8 days (38°C, 100°F)	2.0	0.1	Slightly stained
Citric, 10%	4 months (60°C, 140°F)	1.6	0.8	Slightly softened
Citric, 10%	1 year	1.4	0.6	Unchanged
•Citric, 60%	4 months (60°C, 140°F)	—	—	Surface attacked
Fluosilicic, 10%	2 months	4.5	1.2	Unchanged
Fluosilicic, 28%	2 months	4.7	3.6	Unchanged
Formic, 3%	20 days	—	—	Unchanged
Hydrochloric, 10%	1 year	0.9	0.5	Surface slightly attacked
•Hydrofluoric, 10%	1 month	10.3	5.5	Slightly swollen and softened
•Hydrofluoric, 48%		—	—	Dissolved
Lactic, 50%	2 days	1.6	0.5	Unchanged
•Nitric, 10%	8 months	—	—	Decomposed
Oleic	1 year	2.3	1.5	Unchanged
Phosphoric, 30%	2 months	1.3	0.8	Unchanged
Phosphoric, 50%	2 months	1.6	0.8	Unchanged
•Phosphoric, 75%	2 months	—	—	Partially decomposed
Pyrogallic, 4%	1 week	2.6	1.1	Stained yellow
Stearic	1 week	—	—	Unchanged
Sulfuric, 3%	1 year	1.6	1.0	Slightly discolored
Sulfuric, 10%	1 year	1.5	0.7	Slightly discolored
•Sulfuric, 20%	1 year	0.9	0.3	Slightly softened, surface
attacked •Sulfuric, 30%	1 year	0.4	0.3	Surface attacked
•Sulfuric, 94%		—	—	Disintegrated
Tannic, 10%	4 months (38°C, 100°F)	2.8	1.2	Unchanged
Trichloroacetic, 1%	1 month	3.3	0.5	Unchanged
•Trichloroacetic, 5%	1 month	9.3	3.1	Softened
Alcohols, Monohydric				
n-Amyl	2 days	3.1	3.0	Unchanged
•tert-Amyl	2 days	14.0	11.3	Softened, tacky
•n-Butyl	2 days	6.5	7.2	Swollen
•sec-Butyl	2 days	7.2	10.7	Swollen
•tert-Butyl	2 days	3.6	3.3	Slightly softened
•Diacetone		—	—	Dissolved
•Ethyl (denatured)	2 days	23.0	24.7	Softened
•Ethyl, 50%	1 week	13.4	11.6	Softened
•2-Ethylhexyl	1 week	—	—	Swollen
Hydrocarbons				
Gas, Natural, aromatic-free	1 year	—	—	Showed slight decrease in tensile strength and increase in impact strength
Gas, Natural, 5% aromatic content	23 days	—	—	Showed slight decrease in tensile strength and increase in impact strength

•Indicates that material is generally unsatisfactory for use in contact with Tenite butyrate under the conditions of this test

Chemical Compatibility Chart continued

Reagent	Time Exposed	Percent Increase		Observed Condition
Chemicals		Weight	Thickness	of Plastic
•Isoamyl	2 days	2.0	2.1	Very slightly softened
•Isopropyl	2 days	23.4	25.1	Softened, tacky
•Methyl		—	—	Dissolved
Methyl, 5%	1 year	2.0	1.2	Slightly softened
•n-Propyl	2 days	15.0	4.4	Slightly softened
•Tetrahydrofurfuryl		—	—	Dissolved
Alcohols, Dihydric and Trihydric				
•Diethylene Glycol	2 months	8.2	6.1	Softened
2-Ethyl Hexanediol-1,3	2 days (38°C, 100°F)	—	—	Unchanged
Ethylene Glycol	1 year	4.2	2.1	Unchanged
Glycerin	1 year	0.0	0.4	Unchanged
Propylene Glycol	2 days	0.4	0.0	Unchanged
•Triethylene Glycol	2 months	8.6	6.7	Softened
Bases				
•Ammonium Hydroxide, 10%				
	2 months	21.9	12.9	Softened
Calcium Hydroxide, saturated solution				
	1 week	0.7	0.7	Unchanged
Sodium Hydroxide, 1%	1 year	1.0	0.6	Unchanged
•Sodium Hydroxide, 10%	8 months	3.2	2.2	Brittle
Trimethylbenzyl Ammonium Hydroxide, 5%				
	17 days	1.1	0.0	Unchanged
Esters				
•n-Butyl Acetate		—	—	Dissolved
•sec-Butyl Acetate		—	—	Dissolved
Di-2-Ethylhexyl Adipate	1 year	0.9	0.3	Unchanged
Di-2-Ethylhexyl Phthalate	1 month (50°C, 122°F)			Small gain
•Ethyl Acetate		—	—	Dissolved
•Ethyl Lactate		—	—	Dissolved
•Ethyl Propionate		—	—	Dissolved
•Ethylene Glycol Monoethyl Ether Acetate		—	—	Dissolved
•Ethylene Glycol Monomethyl Ether Acetate		—	—	Dissolved
•Isoamyl Acetate		—	—	Dissolved
•Isobutyl Acetate		—	—	Dissolved
•Isopropyl Acetate		—	—	Dissolved
•Methyl Acetate		—	—	Dissolved
•n-Propyl Acetate		—	—	Dissolved
Ethers				
•Dichloro Diethyl Ether		—	—	Dissolved
•Diethyl Ether	2 days	46.0	50.0	Considerably swollen
Di-Isopropyl Ether	2 days	0.8	1.1	Unchanged
Ether-Alcohols				
•Ethylene Glycol Monoethyl Ether		—	—	Dissolved
•Ethylene Glycol Monomethyl Ether		—	—	Dissolved
•Toluene	2 days	39.3	54.9	Softened
•Xylene	1 week	41.5	33.2	Softened
Hydrocarbons, Halogenated				
•Carbon Tetrachloride	2 days	14.8	6.8	Surface slightly softened
•Chlorobenzene		—	—	Dissolved
•Chlorobromomethane		—	—	Dissolved
•Chloroform		—	—	Dissolved
•o-Dichlorobenzene	3 days	—	—	Softened and swollen
•p-Dichlorobenzene	3 days	11.1	11.7	Swollen

Chemical Compatibility Chart continued

Reagent	Time Exposed	Percent Increase		Observed Condition
Chemicals		Weight	Thickness	of Plastic
•Ethylene Chloride		—	—	Dissolved
•Methylene Chloride		—	—	Dissolved
•Propylene Chloride		—	—	Dissolved
•s-Tetrabromoethane	3 days	—	—	Softened, swollen, and tacky
•Tetrachloroethane		—	—	Dissolved
•Tetrachloroethylene	12 days	—	—	Badly swollen
•Trichloroethylene	1 day	—	—	Badly swollen
Ketones				
•Acetone		—	—	Dissolved
•Cyclohexanone		—	—	Dissolved
•Di-Isopropyl Ketone		—	—	Dissolved
•Methyl Ethyl Ketone		—	—	Dissolved
•Methyl n-Butyl Ketone		—	—	Dissolved
•Methyl Isobutyl Ketone		—	—	Dissolved
•Phorone		—	—	Dissolved
Salts				
Aluminum Acetate, Basic, 33% water slurry				
	2 months	1.8	0.6	Unchanged
Aluminum Chloride, 10% saturated solution	2 months	1.5	0.7	Unchanged Aluminum Chloride,
	2 months	0.1	0.0	Unchanged
Aluminum Sulfate, solid	3 months	1.7	1.1	Unchanged
Ammonium Bifluoride, saturated solution	1 month	2.3	—	Slightly bleached
Ammonium Chloride, saturated solution	1 month	2.1	0.8	Unchanged Ammonium Nitrate,
solid	1 week	0.2	0.2	Unchanged Ammonium Nitrate, 10%
1 week	1.7	0.8	Unchanged	Ammonium Sulfate, solid 1 year
0.1	0.6	Unchanged		
Ammonium Sulfate, 10%	1 year	1.3	0.5	Unchanged
Calcium Chloride, 2.5%	1 year	1.5	0.9	Unchanged
Calcium Chloride, 40%	2 months	0.4	0.0	Unchanged
Calcium Hypochlorite, 6%	1 year	6.0	-3.8	Softened and cracked
Calcium Hypochlorite, 30%	13 days	0.8	0.0	Unchanged
Calcium Phosphate, Monobasic, solid	1 year	1.7	0.5	Unchanged
Calcium Phosphate, Dibasic, solid	1 year	-0.6	0.6	Unchanged
Calcium Phosphate, Tribasic, solid	1 year	-0.6	0.6	Unchanged
Heptane	1 year	1.6	2.5	Unchanged
Hexane	1 week	—	—	Unchanged
Propane, gas	2 months	0.3	0.6	Unchanged
Propane, liquid	2 months	1.4	4.6	Unchanged
Calcium Sulfate (Gypsum), solid	1 year	-0.1	0.6	Unchanged
Copper Sulfate (Cupric), 10%	2 months	1.7	0.6	Unchanged
Copper Sulfate, saturated solution	2 months	1.7	0.9	Unchanged
Cuprous Chloride, solid	1 week	1.5	—	Unchanged
Ferric Ammonium Sulfate, solid	1 week (38°C, 100°F,	0.3	—	Unchanged
80% RH) Ferric Chloride, 5%	2 months	2.0	0.8	Unchanged
Ferric Chloride, 20%	2 months	1.7	1.0	Unchanged
Ferric Chloride, 40%	2 months	1.3	0.4	Unchanged

Chemical Compatibility Chart continued

Reagent	Time Exposed	Percent Increase Weight Thickness	Observed Condition of Plastic
Ferric Chloride, saturated solution	1 month	0.9 0.3	Unchanged
Lithium Bromide, solid	1 week	-0.7 0.0	Unchanged
Lithium Bromide, 50%	1 week	0.0 0.0	Unchanged Magnesium Carbonate,
2.5%	2 days	1.6 1.0	Unchanged
Potassium Aluminum Sulfate (alum), 21%	4 months (38°C, 100°F)	1.9 0.8	Unchanged
Potassium Bromide, 3%	3 days (38°C, 100°F)	1.3 —	Unchanged
Potassium Chloride, solid	1 year	0.1 0.5	Unchanged
Potassium Chloride, 10%	1 year	1.7 0.4	Unchanged Potassium Chrome
Alum, 10%	3 days (38°C, 100°F)	1.3 —	Unchanged
Potassium Cyanide, 10%	2 months	1.4 0.3	Slightly discolored (brown)
Potassium Cyanide, saturated solution	2 months	0.5 0.0	Slightly discolored (brown)
Potassium Ferricyanide, 10%	4 days	— —	Unchanged
Potassium Sulfate, solid	1 year	0.1 0.6	Unchanged
Potassium Sulfate, 10%	1 year	1.4 0.4	Unchanged
Silver Nitrate, 2.5%	2 days	1.5 0.0	Unchanged
Sodium Acetate, 3%	3 days (38°C, 100°F)	1.3 —	Unchanged
Sodium Aluminum Sulfate, solid	1 week	1.6 0.4	Unchanged
Sodium Bicarbonate, 2.5%	2 days	1.7 0.5	Unchanged
Sodium Bisulfate, solid	1 week (38°C, 100°F,	0.1 —	Unchanged
80% RH) Sodium Bisulfate, 1%	3 days (38°C, 100°F)	1.3 —	Unchanged
Sodium Bisulfite, 20%	1 week	2.1 0.8	Unchanged
Sodium Borate, 2.5%	2 days	1.5 0.5	Unchanged
Sodium Carbonate, solid	1 week (38°C, 100°F,	3.9 —	Unchanged
80% RH) Sodium Carbonate, 2.5%	1 year	1.3 0.9	Unchanged Sodium Carbonate,
6% 3 days (38°C, 100°F)	1.2 —	Unchanged	
Sodium Carbonate, 10%	1 year	— —	Unchanged
Sodium Chloride, 2.5%	1 year	— —	Unchanged
Sodium Chloride, 10%	1 year	1.3 0.5	Unchanged
Sodium Chloride, saturated solution	2 months	0.8 0.3	Unchanged
Sodium Chloride, saturated solution	2 months (60°C, 140°F)	0.9 0.9	Unchanged
Sodium Chromate, saturated solution	1 week	0.6 0.2	Unchanged
Sodium Cyanide, 10%	2 months	1.0 0.3	Unchanged
Sodium Cyanide, saturated solution	2 months	-0.2 -0.4	Unchanged
Sodium Ferrocyanide, solid	1 week	0.8 —	Unchanged
Sodium Fluoride, 4%	1 month	2.5 —	Unchanged
Sodium Hypochlorite, 30%	13 days	1.1 -2.1	Unchanged
Sodium Nitrate, solid	2 months	0.1 0.1	Unchanged
Sodium Nitrate, 10%	1 year	1.2 0.4	Unchanged
Sodium Nitrate, saturated solution	2 months	0.9 0.4	Unchanged
Sodium Silicate, solid	2 months	0.3 0.2	Unchanged
Sodium Silicate, saturated solution	2 months	1.2 0.1	Unchanged

Chemical Compatibility Chart continued

Reagent	Time Exposed	Percent Increase		Observed Condition of Plastic
		Weight	Thickness	
Sodium Sulfite, 10%	1 week	2.1	0.9	Unchanged
Sodium Thiosulfate, 20%	13 days	1.1	0.0	Unchanged
Sodium Thiosulfate, 24%	3 days (38°C, 100°F)	1.2	—	Unchanged
Tetra (2-Ethylbutyl) Silicate	1 month (50°C, 122°F)	-0.7	-0.1	Unchanged
Trimethyl Benzyl Ammonium Chloride, 5%	17 days	1.1	0.1	Unchanged
Zinc Chloride (hydrous salt)	1 week	0.5	0.0	Unchanged
Zinc Chloride, saturated solution	1 week	1.4	0.8	Slightly etched
Zinc Oxide, solid	1 week	—	—	Unchanged
Miscellaneous Chemicals and Gases				
Ammoniated Mercury	1 week (60°C, 140°F)	—	—	Unchanged
•Aniline		—	—	Dissolved
•Benzaldehyde		—	—	Dissolved
•Butadiene-1,3, liquid	6 months	19.3	26.4	Swollen and softened
Butadiene-1,3, gas	1 month	2.7	2.3	Unchanged
•Carbon Disulfide	1 week	25.8	1.6	Softened and swollen
•Carbon Disulfide, saturated atmosphere	2 days	17.4	11.8	Warped
•Chlorine, dry	1 week	8.8	2.2	Crazed and brittle
•Chlorine, moist	1 week	7.8	0.1	Crazed and brittle
•Chlorine, saturated solution	1 week	—	—	Considerably softened and swollen
•1,4-Dioxane		—	—	Dissolved
Ethylene Oxide, gas	10 minutes (41°C, 105°F)	—	—	Unchanged
•Ethylene Oxide, gas	1 day	20.9	25.6	Swollen and softened
Formaldehyde, 4%	10 min per day	0.2	—	Unchanged for 5 days
•Formaldehyde, 35%	2 months	13.0	6.7	Swollen and softened
•Furfural		—	—	Dissolved
Hydrogen Peroxide, 3%	1 year	1.7	1.1	Unchanged
Hydrogen Peroxide, 5%	2 days	1.4	1.3	Unchanged
Hydrogen Sulfide, dry	2 months	2.3	0.8	Unchanged
Hydrogen Sulfide, moist	2 months	3.0	1.7	Unchanged
Hydrogen Sulfide, saturated solution	2 months	5.9	2.3	Unchanged
Hydroquinone, 20 g per gallon	1 week	2.4	1.0	Slightly stained yellow
•Methyl Methacrylate Monomer		—	—	Dissolved
•Nitrobenzene		—	—	Dissolved
Ozone, 0.05-0.15 ppm	45 days (outdoors)	—	—	Unchanged
Ozone, 0.7 ppm	45 days (49°C, 120°F)	—	—	Yellowed
•Phenol	1 week	—	—	Decomposed
•Styrene Monomer		—	—	Dissolved
Sulfur, solid	1 week	—	—	Unchanged
•Sulfur Dioxide, dry	2 months	19.4	8.6	Swollen, slightly warped
•Sulfur Dioxide, moist	2 months	31.9	10.2	Considerably swollen and warped
•Sulfur Dioxide, saturated	2 months	23.2	18.1	Swollen and warped

Chemical Compatibility Chart continued

Reagent	Time Exposed	Percent Increase		Observed Condition of Plastic
		Weight	Thickness	
•Sulfur Dioxide in Hydrocarbons		—	—	Dissolved
•Sulfur Dioxide and Hydrocarbon Vapor				
	2 months	19.2	11.5	Swollen
•Titanium Tetrachloride	3 days	—	—	Very brittle
Trinitrotoluene (TNT), water slurry				
	4 weeks	—	—	Stained

Tumor Positioning and Filling the Phantom

The following steps are used to place the rotation ball, position rod and tumor vessel into the phantom. First, loosen the lock bolts on the position rod to insert it through the rotation ball until the rod extends the desired distance, then tighten the lock bolts. Next, position the rotation ball assembly and tumor vessel (or dosimeter) within the phantom. Check to see if the tumor vessel or chamber is in the proper position.

Once you have completed the necessary adjustments to the ball, rod, and tumor vessel, remove the intact assembly from the phantom and fill the phantom with water. Next, insert the assembly back into the phantom and fit the rotation ball lock nut loosely onto the phantom. Rotate the ball until the tumor vessel (or dosimeter) is in the desired position, and tighten the rotation ball lock nut.

Place the pelvis on its back to check for leaks. Small leaks can generally be stopped by tightening the rotation ball lock nut and position rod lock bolt. If you are unable to stop leakage from the rotation ball port, it may be necessary to disassemble the port and check the O-rings to ensure that they are properly seated and lubricated.

Filling the Tumor Vessel

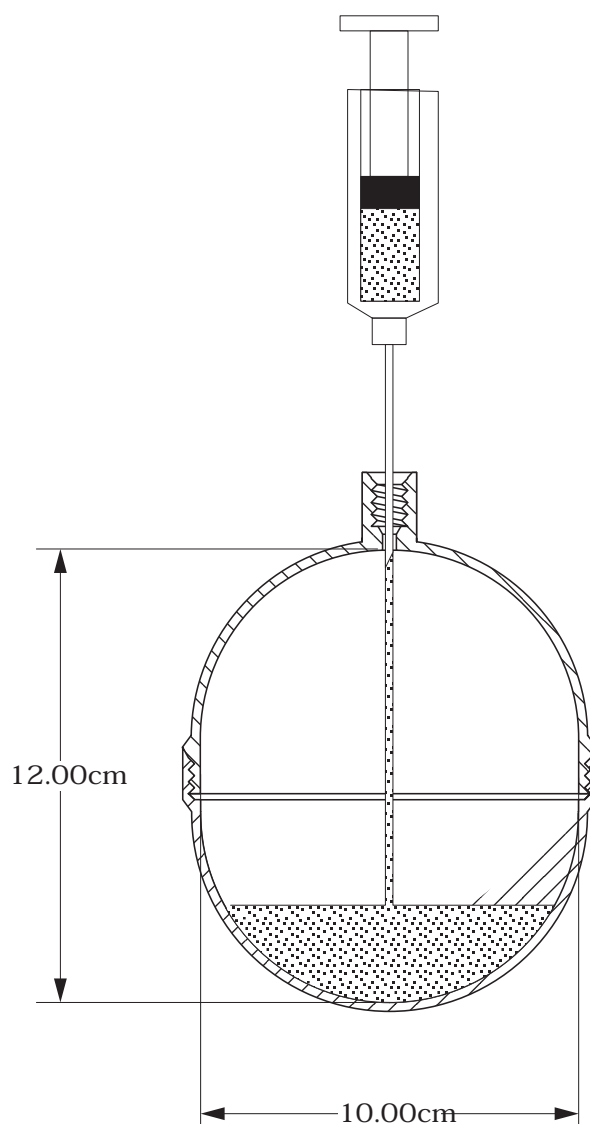


Figure 3

Filling tumor vessel with low viscosity liquids

As shown in Figure 3, the tumor vessel has a small port at the base of the position rod connector. Therefore, you must remove the position rod before adding or removing materials through the port. To fill the tumor vessel with a low-viscosity liquid, you may use a syringe to inject the fluid directly through the port into the tumor vessel. Once the vessel is full, simply rethread the position rod into the connector to close the port.

Filling the tumor vessels with high viscosity materials

A different procedure is used for filling the tumor vessel with high-viscosity materials that cannot be injected through the port. In such cases, it is necessary to unthread the two halves of the tumor vessel and then place the desired material in each half. While the half of the tumor vessel with male threads should be filled completely with material, the female threaded half should be filled only to the beginning of the first thread nearest the rounded end. The two halves of the tumor vessel may then be threaded together. Please note that you may set a tumor-simulation mark by inserting an object such as a catheter section into the gel before threading the two halves together.

Dosimetry Gels

The tumor vessel is molded out of Barex plastic. This plastic is oxygen impermeable. This vessel can be used with a variety of gels. If you prefer not to make your own gel, vessels can be filled by MGS Research Inc. in Guilford CT. 203-453-8679

Preparing the Ferrous Agarose Gel

You may use the following seven-step "recipe" to prepare 50ml of ferrous agarose gel:

WARNING: Beware of any impurities, for example dirt, etc. Impurities will cause the gel to change color prior to irradiation.

- (1) Bring 45ml of triply de-ionized water to a boil.
- (2) Measure 1 gram of agar.
- (3) Add the agar to the boiling water and heat the mixture while stirring continuously for 4 to 5 minutes.
- (4) Remove the mixture from heat and continue stirring for 3 to 4 minutes.
- (5) While the agar mixture is being stirred (step 4), combine 5ml of a ferrous ammonium sulfate solution and 5ml of xylenol orange solution. Directions for preparing these solutions are provided below.
- (6) Once the stirring in Step 4 is completed, add the ferrous ammonium sulfate xylenol orange solution to the agar mixture and stir for an additional 8 to 10 minutes.
- (7) Pour the resulting gel into the two halves of the tumor vessel.

Original Chemicals Needed:

2 x 10⁻⁴M Ferrous Ions (Fe⁺⁺)
 1.5 x 10⁻⁴M Xylenol Orange
 5 x 10⁻²N Sulfuric Acid (H₂SO₄)
 1.0% Agar

Preparing Stock Solutions:

Xylenol Orange

- (1) Measure 0.124 grams of xylenol orange.
- (2) Add 100ml of triply de-ionized H₂O.

Ferrous Ammonium Sulfate

- (1) Measure .0784 grams of Fe⁺⁺.
- (2) Add 25ml of H₂SO₄ (5x 10⁻² Normal).
- (3) Add 75ml of triply de-ionized water.

Chemical Supply Source:

Fisher Scientific Headquarters, 711 Forbes Avenue, Pittsburgh, PA 15219
 412-562-8300

Ferrous Ammonium Sulfate crystal [Fe(NH₄)₂(SO₄)₂·6H₂O] (Cat.# I77-500)
 Xylenol Orange Tetrasodium Salt (Cat.# X15-5)

Sigma Chemical Co., St. Louis, MO
 314-771-5750

Purified Agar (Stock# A-7049)
 Ferrous Ammonium Sulfate [Fe(NH₄)₂(SO₄)₂·6H₂O]
 Xylenol Orange (Stock# X-3500 Tetrasodium Salt)

Radiation Chambers and Probes

The rotation ball will hold 1/2" diameter tubing. Water sealed chambers mounted on 1/2" diameter tubes with adequate length and end clearance may be used in this phantom. In some cases custom modifications can be made to allow other sizes.

WARRANTY

THE PHANTOM LABORATORY INCORPORATED ("Seller") warrants that this product shall remain in good working order and free of all material defects for a period of one (1) year following the date of purchase. If, prior to the expiration of the one (1) year warranty period, the product becomes defective, Buyer shall return the product to the Seller at:

By Truck	By Mail
The Phantom Laboratory, Incorporated	The Phantom Laboratory, Incorporated
2727 State Route 29	PO Box 511
Greenwich, NY 12834	Salem, NY 12865-0511

Seller shall, at Seller's sole option, repair or replace the defective product. The Warranty does not cover damage to the product resulting from accident or misuse.

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